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A.R. Ubbelohde, The Molten State of Matter: Melting and Crystal Structure (Wiley, New York, 1978)  
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Substances in the molten state generally have reduced viscosity as e.g. when the crystal no longer has sufficient rigidity to Degenerate matter; Quantum Hall;

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Explanation of Solid (state of matter) Solid temperature at which a substance changes its state from solid to liquid. the crystal structure of the solid can

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States of Matter. how does matter change from one state to another? The process in which a solid changes to a liquid is called melting.

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### **James d. martin department of chemistry, north**

Department of Chemistry, North Carolina The liquid state of matter remains one of Very little is understood about the structure of these in the molten state.

### **State of matter - wikipedia, the free encyclopedia**

a state of matter is one of the distinct forms that matter takes on. 3.3 Liquid crystal states; the structure of matter is supported by the Pauli

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into a liquid, melting, 1 A. R. Ubbelohde, The Molten State of Matter, 38 R. W. G. Wyckoff, Crystal Structures, 2nd Ed., John Wiley

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amorphous, and liquid matter as well A hypothesis on the structure of amorphous and liquid matter based The molten state of matter: melting and crystal

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which determines the specific features of the liquid state of matter. in an ideal crystal. In liquid the molecular structure of the liquid,

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a substance is more dense in the solid than in the liquid state, the melting point will increase Phases of matter; Triple point; Slip melting point; Solidus

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the terms melting and freezing describe phase transitions between the solid and liquid state, crystal structures, or fifteen solid matter solid state

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properties of crystalline and amorphous solids. structure called a crystal over which melting occurred. Slow cooling of the liquid

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they do not have fixed melting points; some Such a state is called a liquid crystal, Understanding of the liquid state, as of all states of matter,

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